

MEMBRANE CHEMICALS



Presented by: Majid Karami

- an book of the control of the contro
- 🖈 Water Shortage is becoming a worldwide problem
 - ★ One of the main solutions to this problem is Desalination of seawater or brackish water
 - ★ Desalination of seawater accounts for a worldwide water production of 24.5 million m3/day and Iran's share is about 505,000 m3/day (2009)
 - ★ Increasing the cost of transferring water and dec the desalination cost, has caused this technology more popular



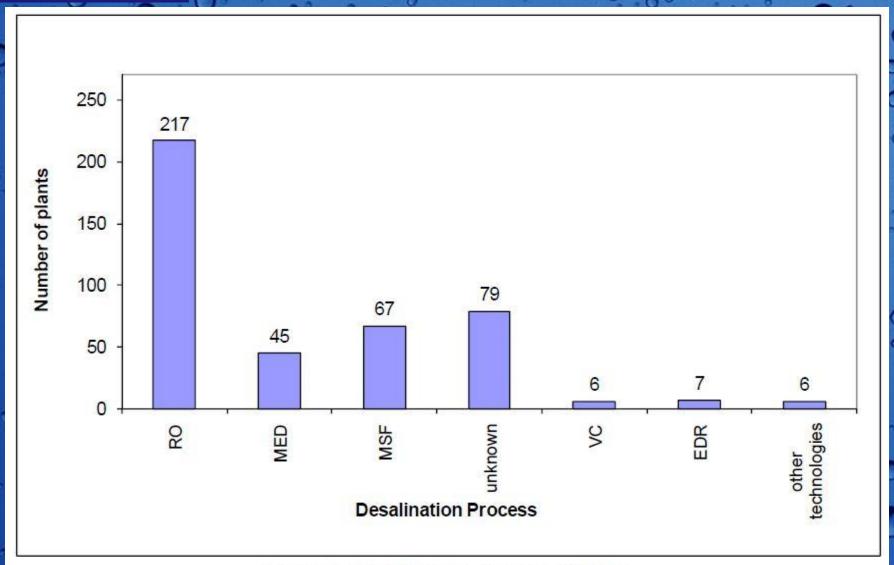


Figure 1. Desalination processes in Iran

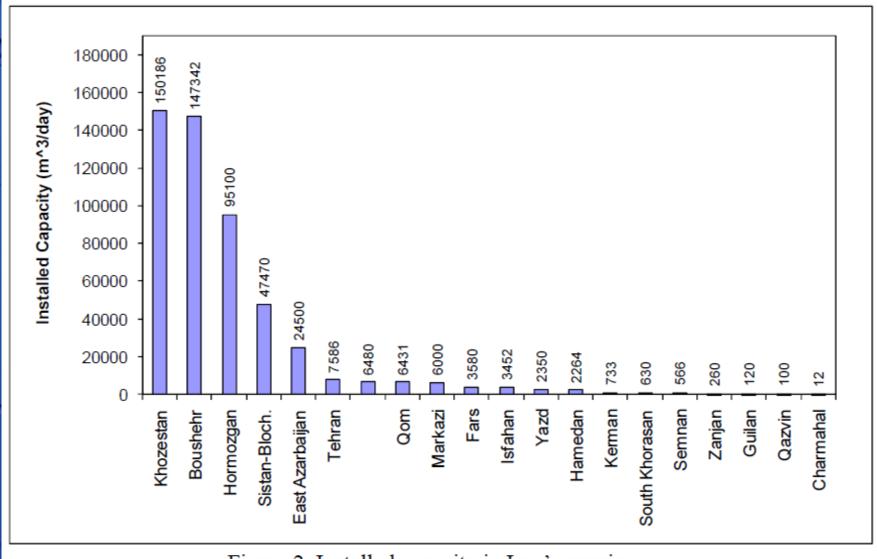


Figure 2. Installed capacity in Iran's provinces

ODesalination technology like any other technology ha

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- some issues and problems such as:
- ★ Fouling a
- ★ Scaling
- ★ Corrosion

Which causes:

- ★ A decrease in operational efficiency
- ★ Negative effects on the environment
- ★ Higher cost of water production in some areas



- Fouling & Sealing
- Fouling includes the accumulation of lighted of layers on the membrane and feed Spacer surface, including scaling of such as:
 - ★ Inorganic colloids: iron flocs, silica, œlay, silt
 - * Organic colloids: oil, organic polymers, microorganisms



Scaling will refer to the precipitation and desorption within the system of sparingly soluble salts such as:

★ Calcium carbonate, barium sulfate, calcium sulfate, strontium sulfate, and calcium fluoride

Pretreatment	CaCO ₃	CaSO ₄	BaSO ₄	SrSO ₄	CaF ₂	SiO ₂	SDI	Fe	Al	Bacteria	Oxid. agents	Org. matter
Acid addition	•							0				
Scale inhibitor antifoulant	0	•	•	•	•	0		0				
Softening with IX	•	•	•	•	•							
Dealkalization with IX	0	0	0	0	0							
Lime softening	0	0	0	0	0	0	0	0				0
Preventive cleaning	0					0	0	0	0	0		0
Adjustment of operation parameter	0	0	0	0	0	•						
Media filtration						0	0	0	0			
Oxidation filtration							0	•				
In-line coagulation							0	0	0			0
Coagulation-flocculation						0	•	0	0			•
Microfiltration/Ultrafiltration						•	•	0	0	0		•
Cartridge filtration						0	0	0	0	0		
Chlorination										•		
Dechlorination											•	
Shock treatment										0		
Preventive biocidal treatment										0		
GAC filtration										0	•	•

		38.9	100		0	10 - 0 1	0	() one ()
Cleaner	0.1 wt % NaOH	0.1 wt % NaOH with 1.0 wt % Na ₄ EDTA	0.1 wt % NaOH with 0.025 wt % Na-DDS	0.2 wt % HCI	2% citric acid	1.0 wt % Na ₂ S ₂ O ₄	0.5 wt % H ₃ PO ₄	1.0 wt % NH₂SO₃H
	NaOn	N44EDTA	เทล-กกว	псі	aciu	IVa232U4	H3PU4	พที่ของสา
Carbonate scales (e.g., CaCO ₃)		e 	-	Preferred	Alternative	Optimal	Alternative	
Sulfate scales (CaSO ₄ , BaSO ₄)		OK	(F) No.	N . V A			62 26	20 10
Metal/oxides hydroxides (e.g., iron)		2 2			Alternative	Preferred	Alternative	Alternative
Fluoride scales		OK						,
Phosphate scales				Preferred				
Inorganic colloids (silt)		2 2	Preferred	-		=	8 8	5-4
Silica	Alternative	Alternative	Preferred	8 8		= 0	12-0	5-0
Biofilms	Preferred	Alternative	Preferred	\$ \$		(-	<u> </u>	# - #
Organic	Preferred	Alternative	Preferred	8 9		s=0	12-8	

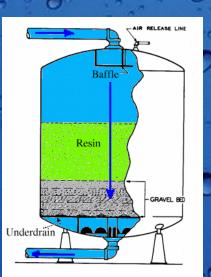
Scale Control Methods

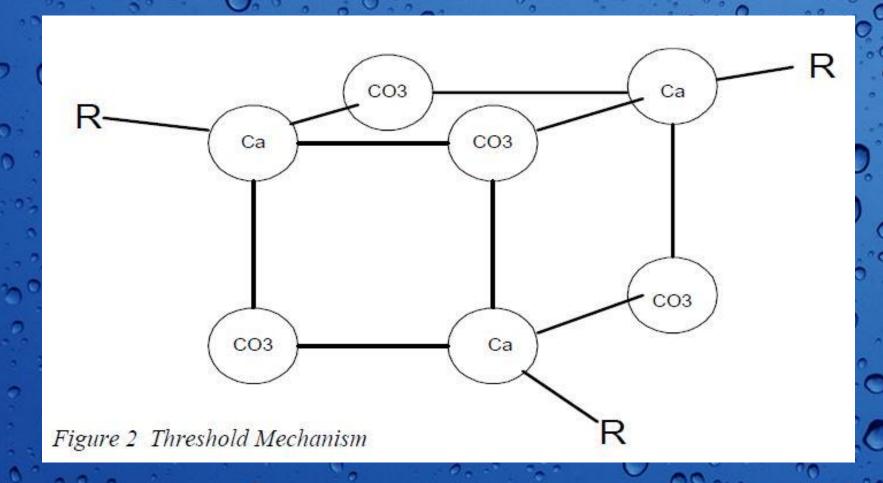
- (1) Acid Addition:
 - ★ Effective in preventing the precipitation of CaCO3
 - ★ Ineffective in preventing other types of scale such as
 - CaSO4
 - ★ Corrosivity of the acid
 - (2) Softening:
 - ★ The main disadvantage of softening is cost (\$

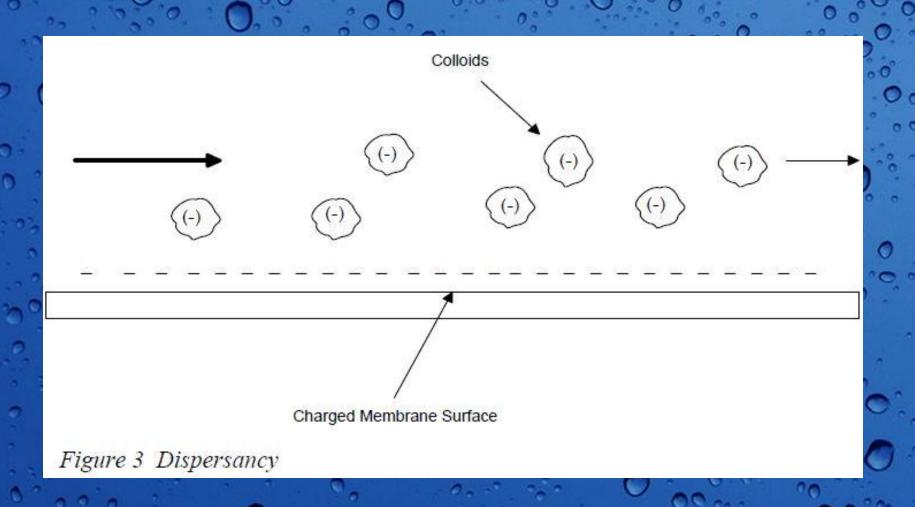
$$Ca^{+2} + 2NaZ \implies 2Na^+ + CaZ_2$$

$$Mg^{+2} + 2NaZ \implies 2Na^+ + MgZ_2$$

(3) Antiscalant Addition







DEQUEST®	CaCO₃	CaSO₄	BaSO₄	CaPO₄	CaF₂	SrSO₄	Silica	Metal Oxides
SPE 0001	✓	•		•	✓	8).		•
SPE 0106	1	✓			√			0
SPE 0107	✓	O	0		1	0	0	✓
SPE 0108	0		✓	✓	0	✓	✓	✓
SPE 0109		0	✓	✓		✓	✓	✓
SPE 0109 POT		0	1	1		✓	√	✓
SPE 0111	✓	Ō		0	√			0
SPE 0112	✓	0			✓		ż	0
SPE 0125	✓	•			✓	3		0

- ✓ Best efficacy
- Good efficacy

Antiscalant Addition

- ODosage rates depends to:
 - ★ Water Analysis
 - ★ Flow rate
 - ★ Recovery
 - ★ Water source
 - \star pH
 - ★ Temperature
 - ★ Membrane type & RO configuration
 - ★ Typical dosage rates are 1-6 mg/lit

Recovery:

★ As the recovery rate increases the concentration of dissolved salts in the reject water stream increases and leading to salt precipitation and scale formation

Dosage calculation:

Dosage (ppm) =
$$\frac{V_1 \times S.G \times 1000}{Q \times V_2} \times Dosing Rate$$

 V_1 = volume of Antiscalant (Lit)

S.G= Specific Gravity of Antiscalant

Q= Feed Flow Rate (m³/hr)

 V_2 = volume of water in dosing tank (Lit)

Dosing Rate (Lit/hr)

★ CoRoLa-T: software for selection of the most appropriate antiscalant for a specific system





Gilles Delaisse @SSQL03\SDATA01						Sign (
D Adioin	G	eneral	Spe	cification	Water analysis	Result
Disecunity Projects		Brine Sat	uration Data		Scaling calculation options	
Projects Overview	Name	Value	Treatment	⊕ pH dhangs	O system recovery change	
Information	L91	1.55				
	18082	0.61			2	
	04904	0.53		Acid Type:	H2804	Sec. 1
	Ba504	0.00		8	A15	

Acidified pH:

Product recommendation!

- Treat feed water with: SPE 0001 at 1.0 ppm dosage
 The daily requirement of SPE 0001: 0.104 kg/day
 The current feedwater pH: 8.00

SHS04

SICZ

CaF2

CaPO4

Fe

0.54

0.83

20.00

6:04

0.00

Copy Project

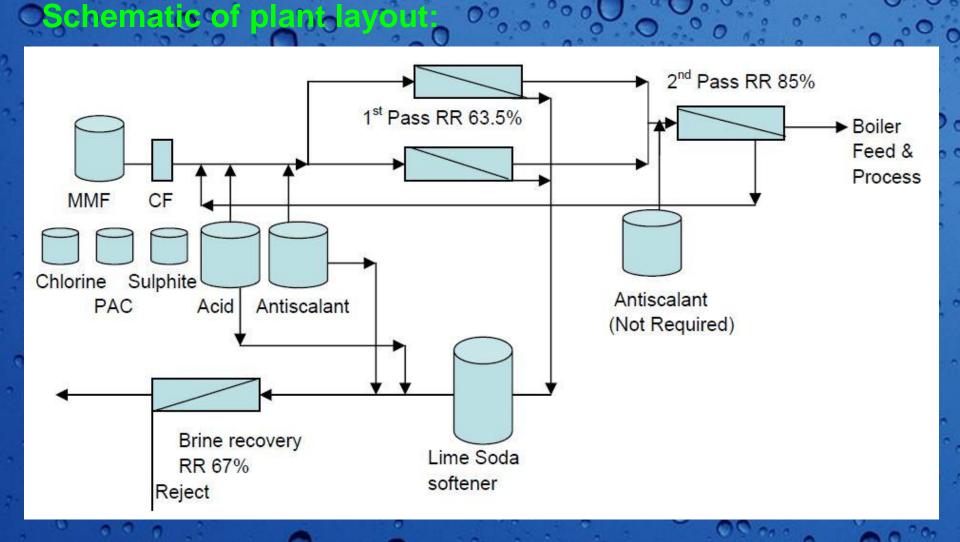
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Continue

- A chemical manufacturing company in learn
 - ★ 7,000 m3/day two pass brackish water RQ plant
 - ★ With brine recovery RO (BRRO).
 - ★ First pass with two identical skids (A&B)
 - ★ Each skid is two stages in 20:10 array
 - ★ Each pressure vessel having 6 membrane elements
 - ★ Both skids (A & B) have a design permeate flow of 143 m3/hr at 63.5% recovery

Case Stric

- ★ Total first pass production of 286 m3/hr
- ★ Pretreatment for the first pass: aluminium salt coagulant, hydrochloric acid injection, chlorination, multimedia filtration, phosphonate based antiscalant injection, SMBS injection, cartridge filteration
- ★ Second pass is designed to operate at 85% recovery



- ODescription:
 - ★ Raw water derives from underground wells.
 - ★ Chemical water analysis (as ions): calclum of 785 mg/l, sulphate at 2149 mg/l, bicarbonate of 144 mg/l, with pH 7.5 and TDS 4745 mg/l

Case Stid

- ★ Scaling prediction: calcium carbonate, calcium sulphate and barium sulphate salts all exceeded their solubility limits
- ★ Standard anttiscalant could inhibit calcium carbonate and barium sulphate but not calcium sulphate at these high values

- OProblem: 0
 - ★ The first pass trains A & B of the plant historically suffered from calcium sulphate deposition

Case Stig

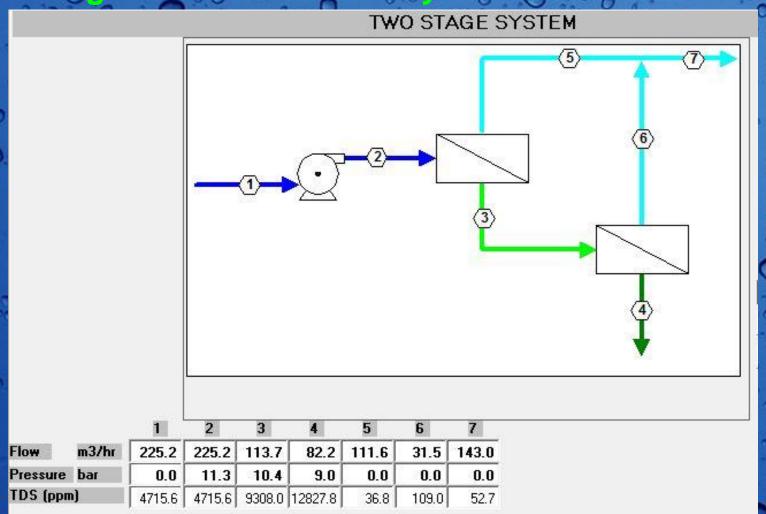
- ★ Operators decided to reduce recovery rates to 489 against a design specification of 63.5%
- ★ Even at lower recovery rates the membranes fouled after 4 5 weeks of operation with feed pressure increasing from 10 to 12.5 bar
- ★ Cleaning was carried out when ∆P reached 6.5 bar by circulating a solution of Ethylene Diamine Tetra Acetic acid (EDTA) and Sodium TripolyPhosphate (STPP) at high pH

- **Recommendations**
 - ★ In November 2007 the new calcium sulphate specific antiscalant was used and acid dosing was stopped

Case Stid

- * At 4.7 mg/l dosed to the feed water this approch has successfully prevented scale formation
- ★ The recovery rates were increased to 61%
- ★ The plant requiring only one membrane clean in 18 cmonths instead of one every month as previously
- ★ The second pass permeate polisher is operating without the need for antiscalant due to the good quality permeate from the first pass

Olnitial design with 63.5%



Results and cost se

8	Permeate	Feed m³/hr @	Feed m³/hr @	Feed Saving	Feed Saving
	m³/hr	48% Recovery	61% recovery	m³/hr	m³/annum
Skid A	143	298	234	64	560,640
Skid B	143	298	234	64	560,640
TOTAL	286	596	468	128	1,121,280

OResults and cost saving ★ Initial design

Feed pressure	
Concentrate pressure	
Permeate flow	
Pump Feed Flow	

bar	\mathbf{T}	11.3
bar		9.0
m3/hr		143.0
m3/hr		225.2

Recovery ratio, Z	
Pump efficiency, 7	
Motor efficiency, 7	

63.5
83.0
93.0

Power/Stage/Pass, kw	
Total Pumping power, kw	
Pumping specific energy	

		89.8
		89.8
kwhr/m3	•	0.64

OResults and cost saving for

0 0		0 2 ,	_0 0	000
Skid 1 a	Nov-03 Design	Sep-07	Feb-08	Feb-09
Feed Pressure Bar	12.80	12.50	10.70	12.00
% Recovery Rate	64	48	58	61
Feed Flow m3/hr	225.20	283.00	243.00	228.00
Permeate Flow m3/hr	143.00	136.00	140.00	139.00
Pump Energykwhr/m³	0.63	0.85	0.58	0.62
Energy KwHr	141.88	240.55	140.94	141.36
Total Energy/annum KwHr	1225808.64	2078352.00	1217721.60	1221350.40
Pumping Costs/ annum	\$85,806.60	\$145,484.64	\$85,240.51	\$85,494.53

Conclusions:

4150	Sulphate Case Study
Water Saving, m ³ /annum	1,121,280
Energy Saving, KwHr	857,002 (1 skid)
Energy Costs Saving, US\$/annum	\$60,000
Membrane Replacement, US\$/annum	\$39,000
Chemical saving, US\$/annum	\$37,000
Total, US\$/annum	\$136,000

- Conclusions:
 - This Case Study demonstrates that the selection of the correct chemical programme and using species specific

Case Stig

- antiscalant can result in:
- * Optimizing the recovery rate so as to minimize pumping costs
- ★ Maintaining membrane cleanliness and reducing cleaning frequency
- ★ Extending the lifespan of the membranes thereby reducing replacement costs
- ★ Removing the dependency and the cost of dosing large quantities of non specific commodity chemicals which frequently has a positive environmental impact